

# Performing a Titration, Part 2

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8

Fill the other buret with standardized base solution to a point above the 0 mL mark. Record the concentration of the standardized solution.



9

Release some base from the buret into a waste flask so that the top of the liquid is in the calibrated portion of the buret.



**10** Record the volume of the base to the nearest 0.01 mL as your starting point.



**11** Place the flask containing the acid under the base buret. Notice that the tip of the buret extends into the mouth of the flask.



**12** Slowly release base from the buret into the flask while constantly swirling the flask. The pink color should fade with swirling.



**13** Near the end point, add base drop by drop.



**14** The end point is reached when a very light pink color remains after 30 s of swirling.



**15** Record the new volume, and determine the volume of base added.